

# WET OXIDATION TO TREAT HTL WASTEWATER

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# WET WASTES



Manure  
1.4 b t EU



Wastewater  
sludge  
12 m t EU



Food waste  
60 m t EU

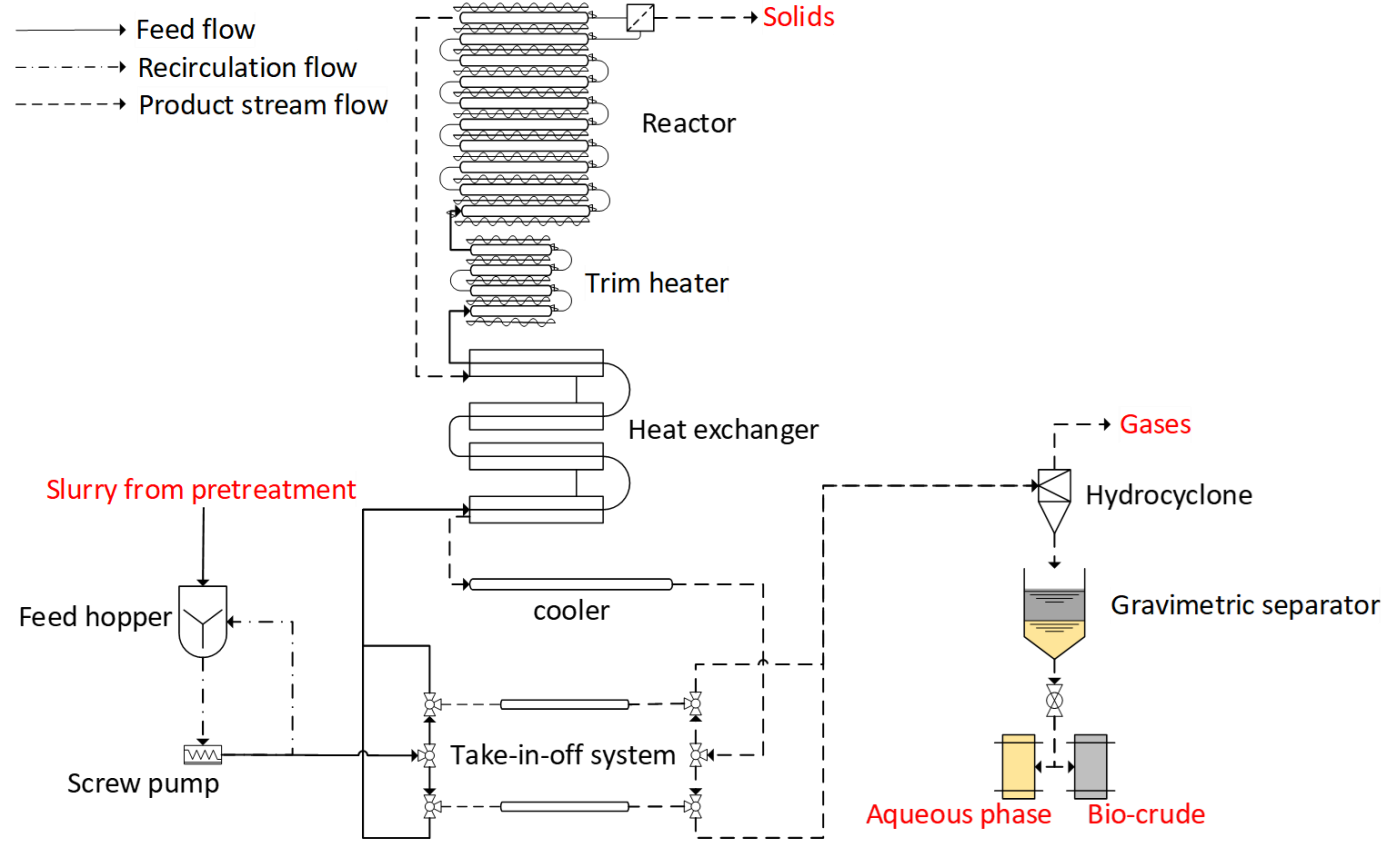


Digestate from AD  
190m t EU

## Commonality:

- High availability
- Less feedstock competition
- High water content
- High inorganic content
- High P content
- High N content
- Presence of micropollutants etc
- Presence of heavy metals

# HTL PILOT PLANT



- 65 L/h slurry flow rate
- 3 kg/h bio-crude production
- Subcritical region (300-350°C)
- Heat exchange >75%
- EROI ~3



# HTL WASTEWATER

What to do with the water?

Temperature	PO4	NH4+	TOC	COD
300°C	4.3	372.5	20,000	50,833
325°C	2.3	635	15,700	42,250
350°C	7.6	720	20,200	51,850

Units mg/L

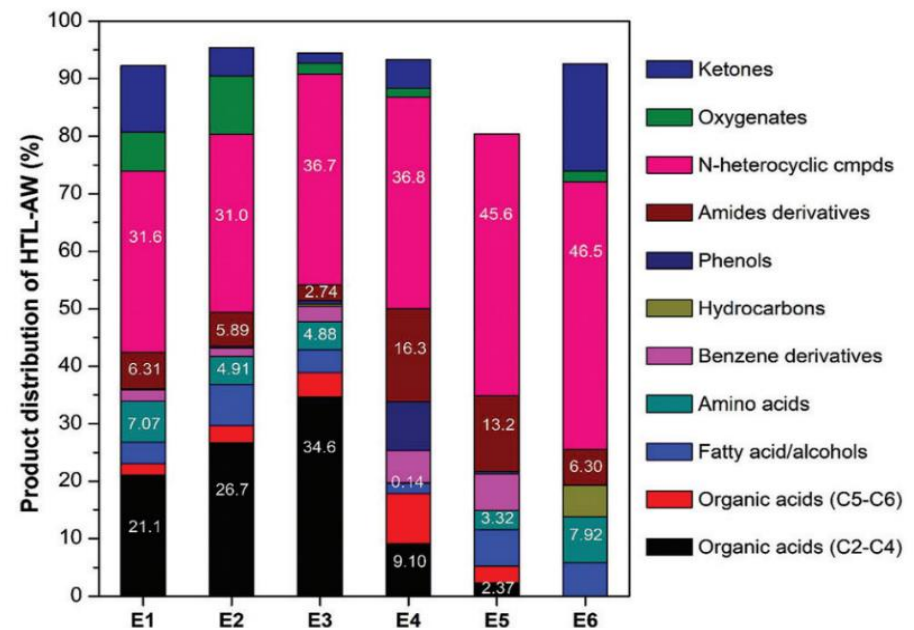
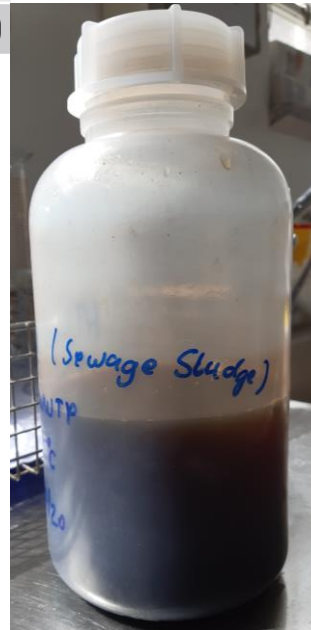
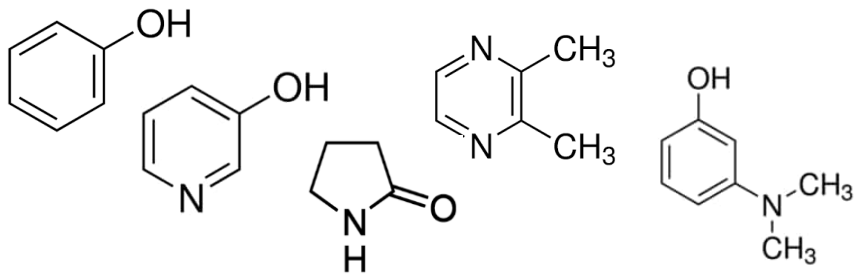


Figure 2 - Shows composition of organic compounds in the HTL wastewater, based on functional groups. E1-E4 shows composition of processed mixed wastewater algae. E5 and E6 are based on *Spirulina* as feedstock to the HTL under different process conditions (Gu et al., 2019)

Green Chem., 2019, 21,

2518-2543

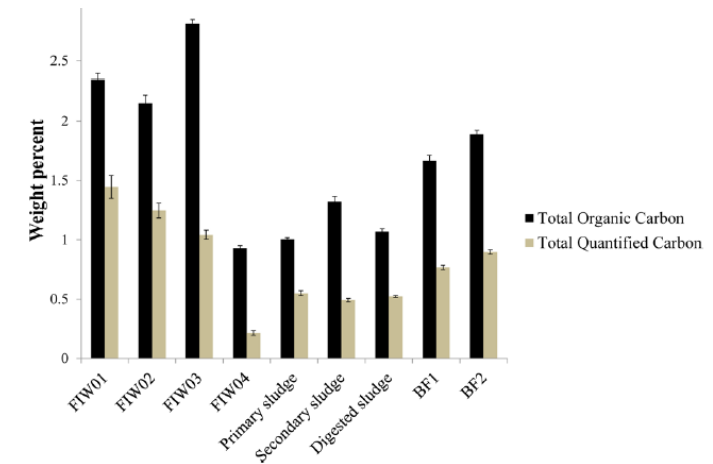


Figure 6. Total organic carbon and quantified carbon of the aqueous byproducts generated from HTL of food industry waste, municipal waste, and biomass grown on wastes: BF1 biomass feedstock grown on corn stover lignin residue; BF2 biomass feedstock grown on municipal waste.

DOI:10.1021/acssuschemeng.6b02367

ACS Sustainable Chem. Eng.2017, 5, 2205-22142208

# BIOLOGICAL TREATMENT

## Results Anaerobic digestion:

- Straw+manure derived water showed no inhibition at 65% COD load
- Sludge derived water inhibition at 15% COD load
- Nitrogen aromatics in sludge water are higher and cause inhibition

## Results WWTP:

No impact of heterotrophic bacteria

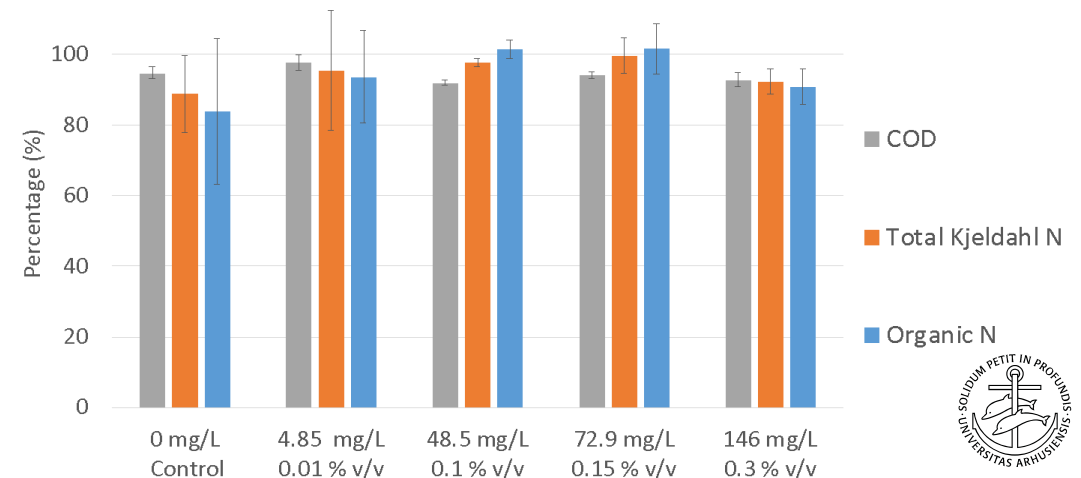
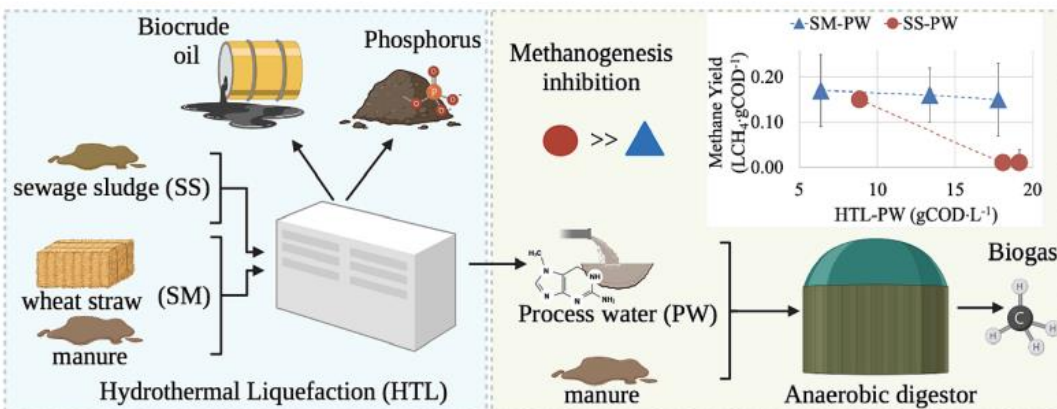
- >92% biodegradability of COD and organic N
- 45% higher denitrification rates with HTL-PW than with influent

Nitrifying bacteria can deal with it

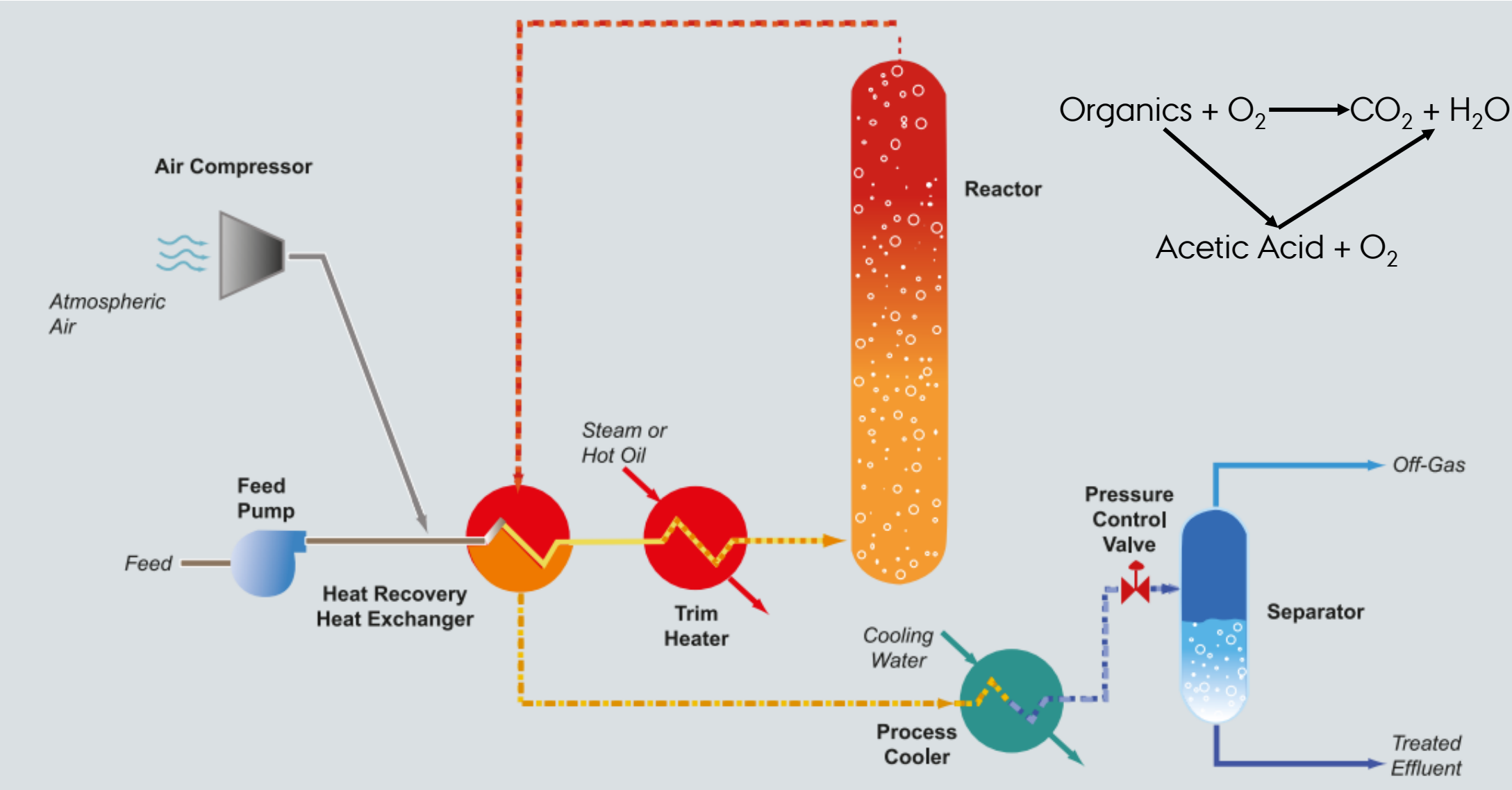
- 8% nitrification inhibition in worst case (shock-load)
- No nitrification inhibition in continuous reactors

**BUT:** Huge load on total inlet COD (~20%)

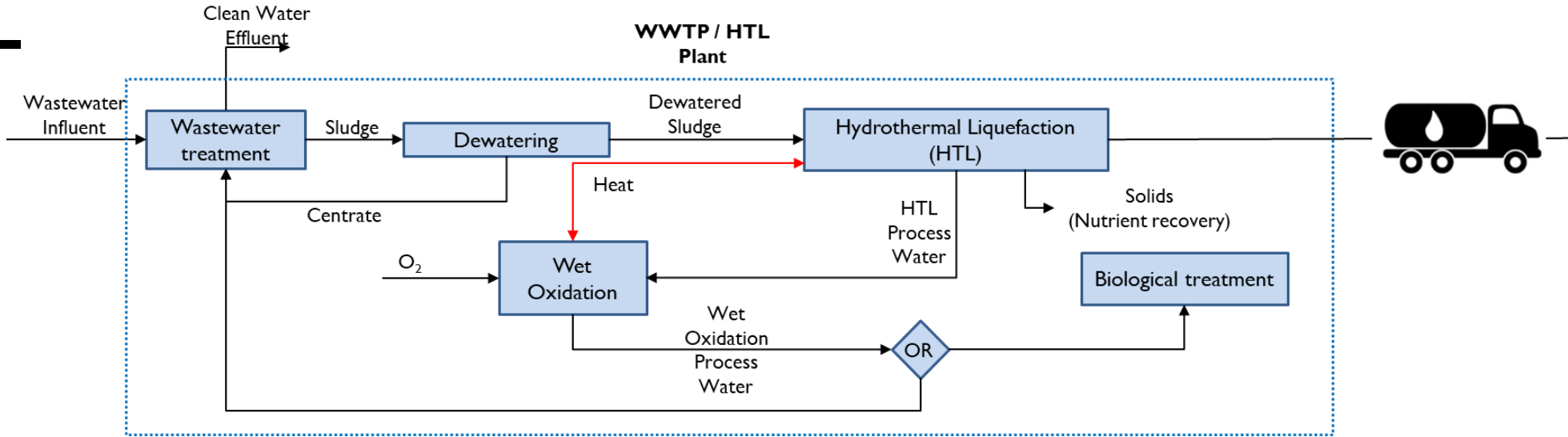
<https://doi.org/10.1016/j.biortech.2024.130559>



# WET AIR OXIDATION PROCESS



# WET OXIDATION



**Continuous Wet Oxidation**

CO<sub>2</sub>

O<sub>2</sub>

Heat

Continuous HTL pilot reactor

Sewage sludge

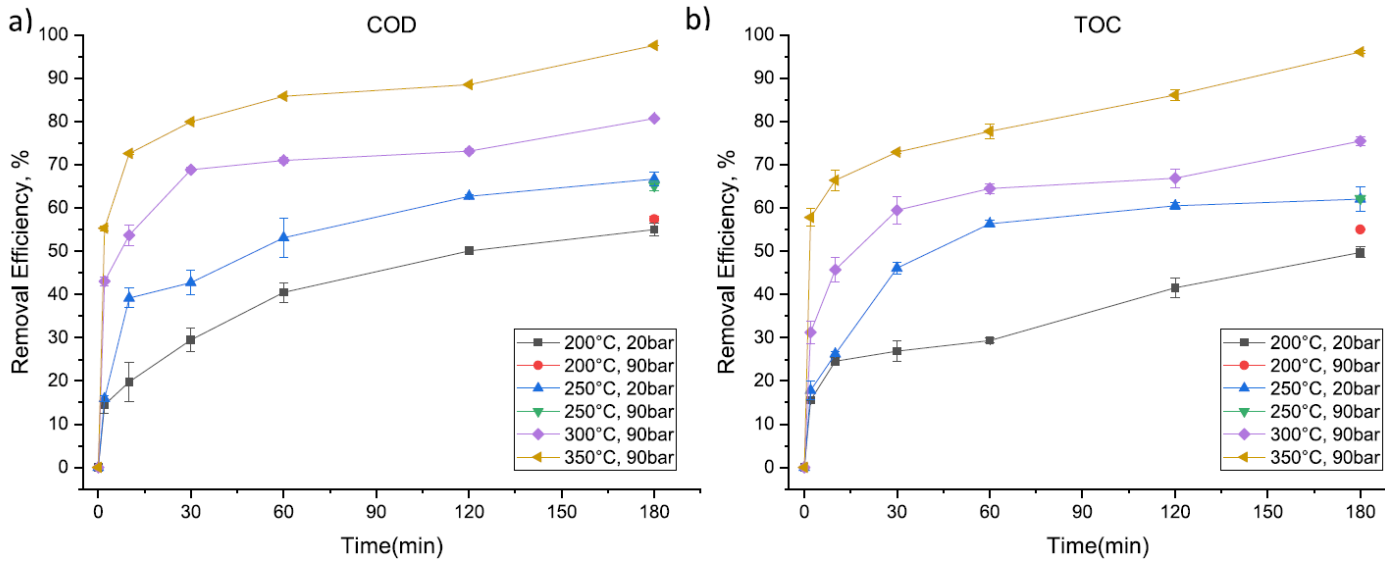
- Gas
- Bio-crude
- Water phase
- Solids

COD: 43 g/L

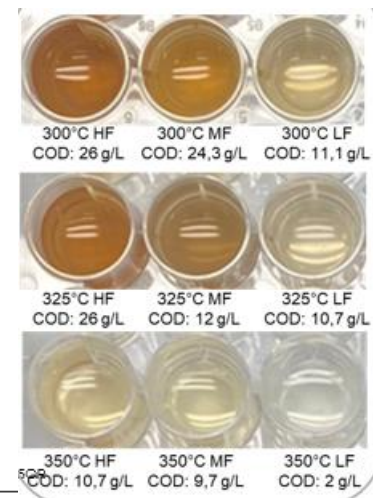
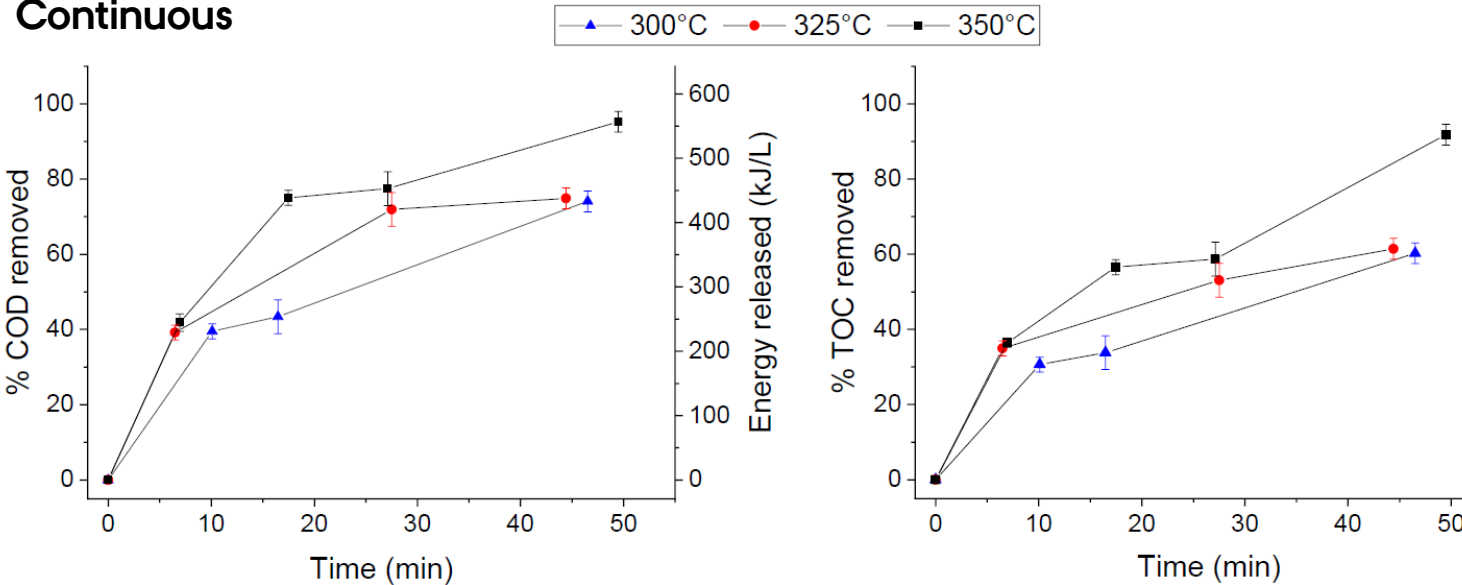
300°C HF COD: 26 g/L	300°C MF COD: 24,3 g/L	300°C LF COD: 11,1 g/L
325°C HF COD: 26 g/L	325°C MF COD: 12 g/L	325°C LF COD: 10,7 g/L
350°C HF COD: 10,7 g/L	350°C MF COD: 9,7 g/L	350°C LF COD: 2 g/L

# WO RESULTS

## Batch



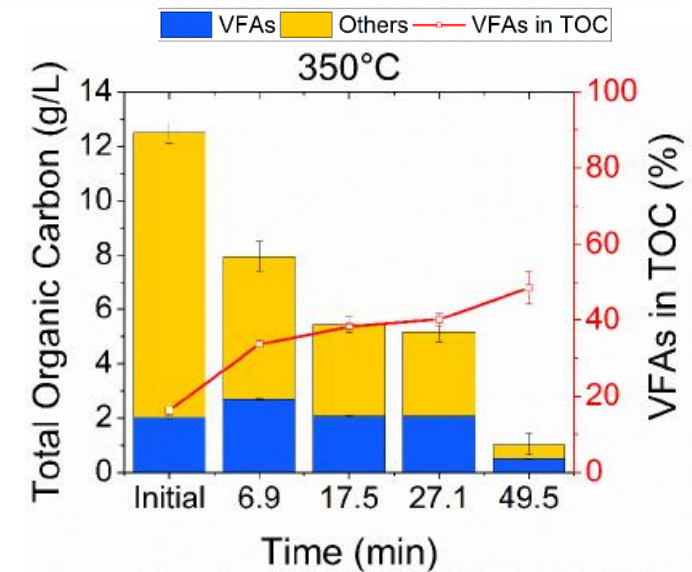
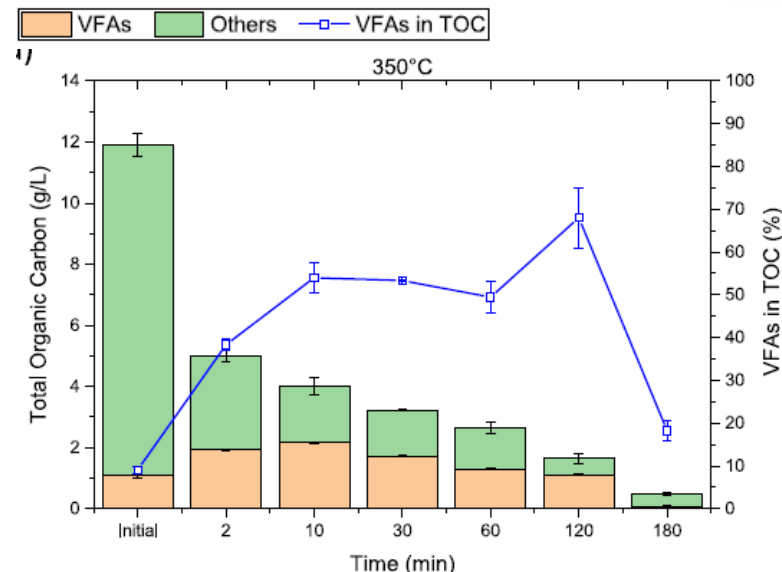
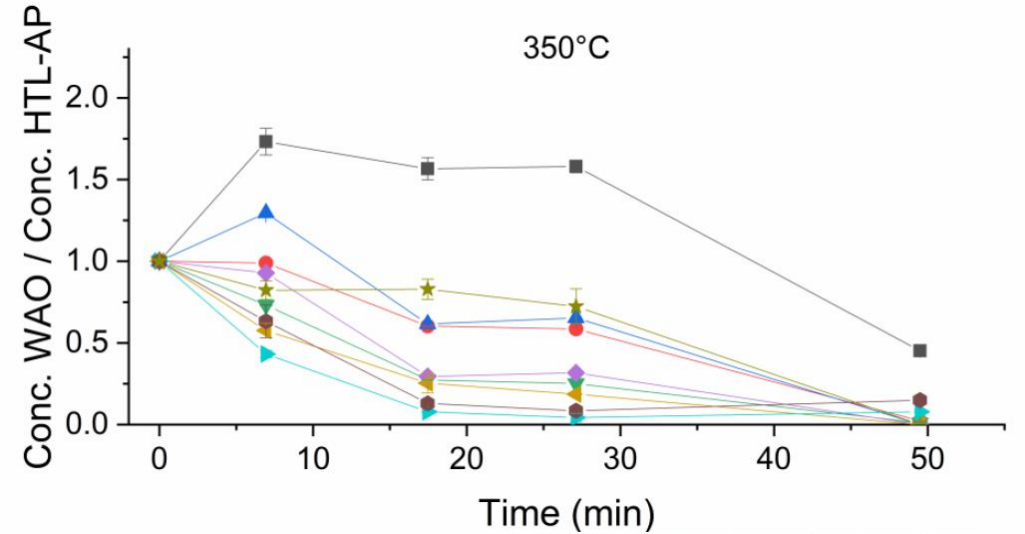
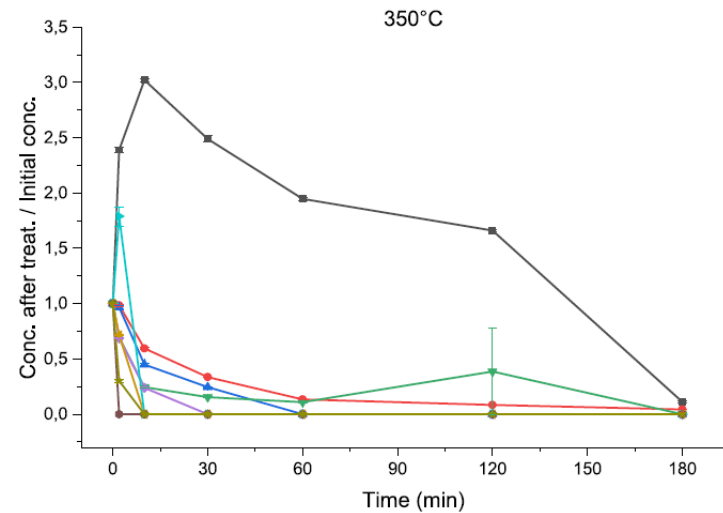
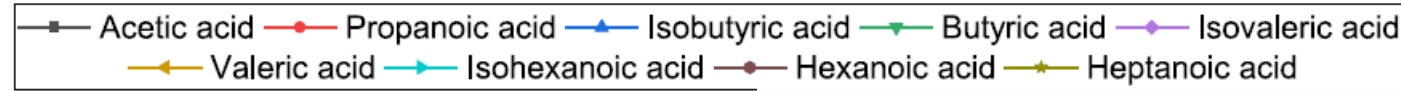
## Continuous





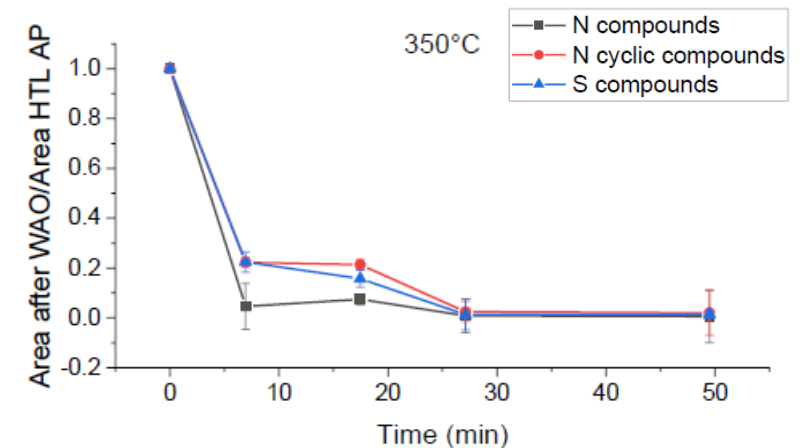
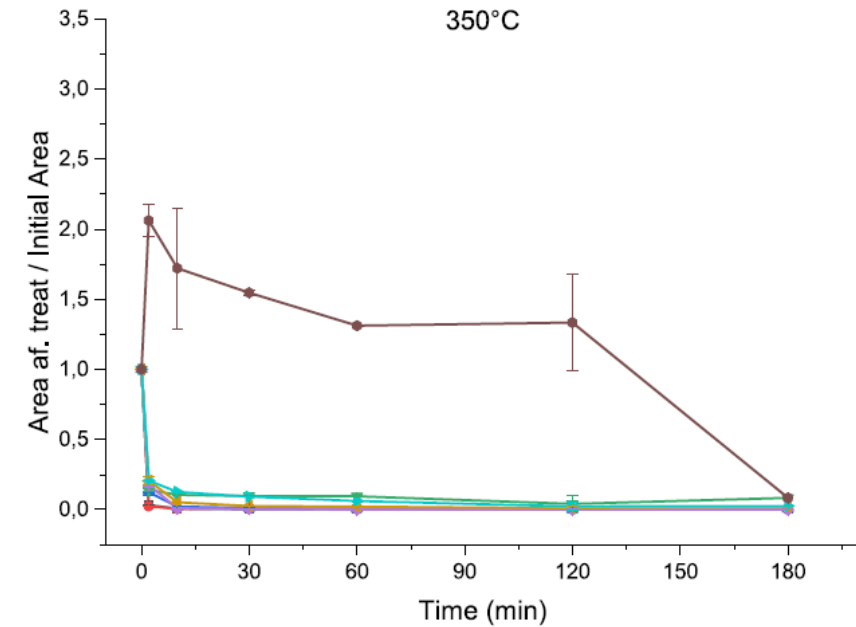
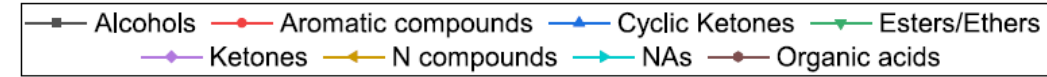
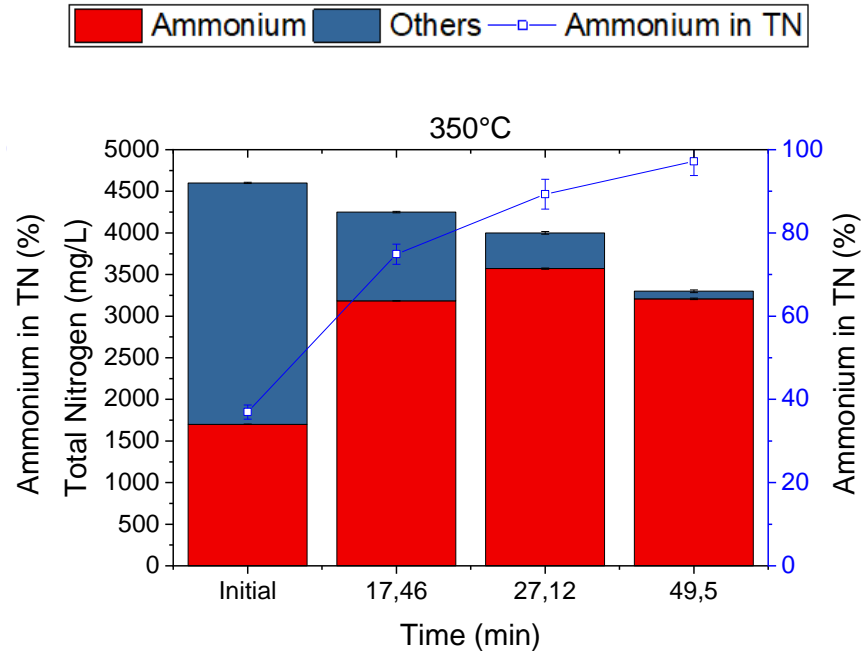
# VFA PRODUCTION

- Stopping the WO process at reasonable residence times is an option
- VFA share is highest
- Opens up possibilities of acetic acid recovery



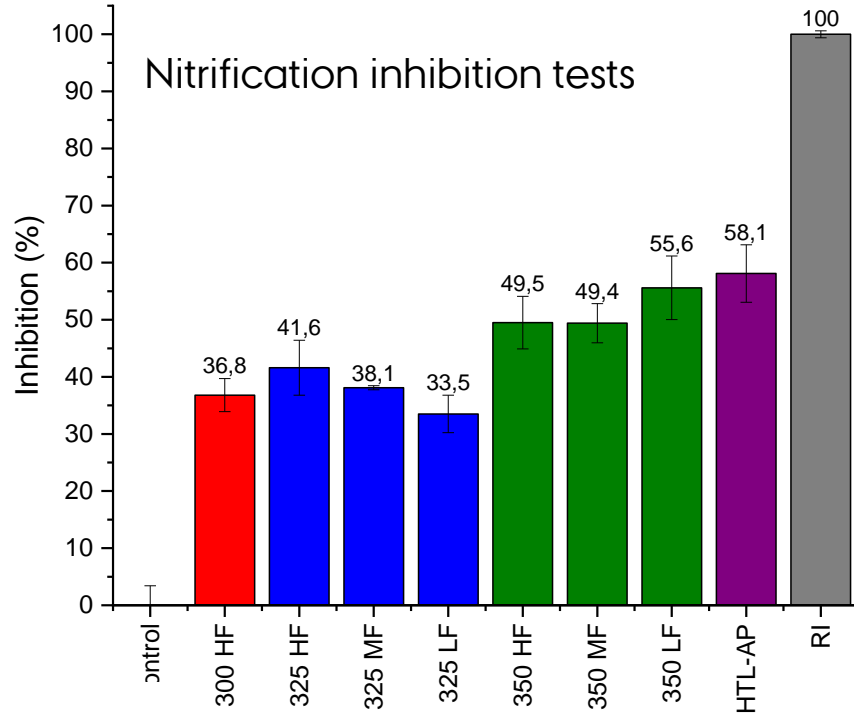
# NITROGEN IN WO

- GC-MS analysis shows fast and almost full degradation of most compound classes
- Organic-N compounds are efficiently converted to NH<sub>4</sub>
- Recovery of NH<sub>4</sub> is attractive

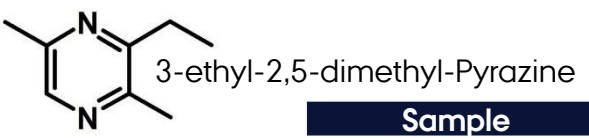
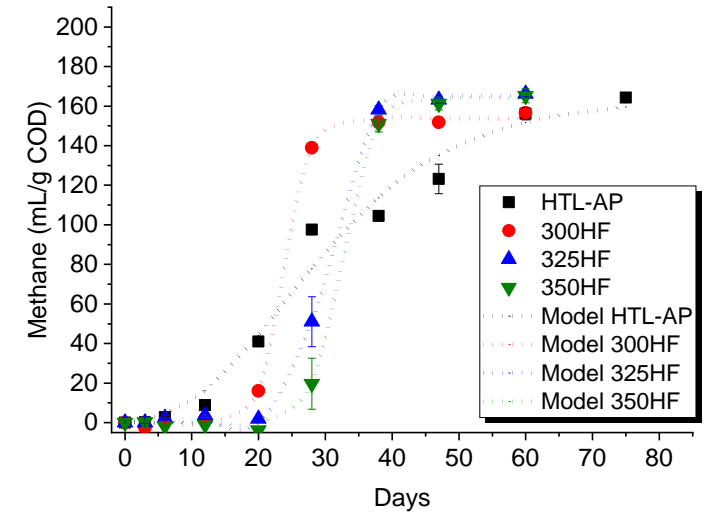
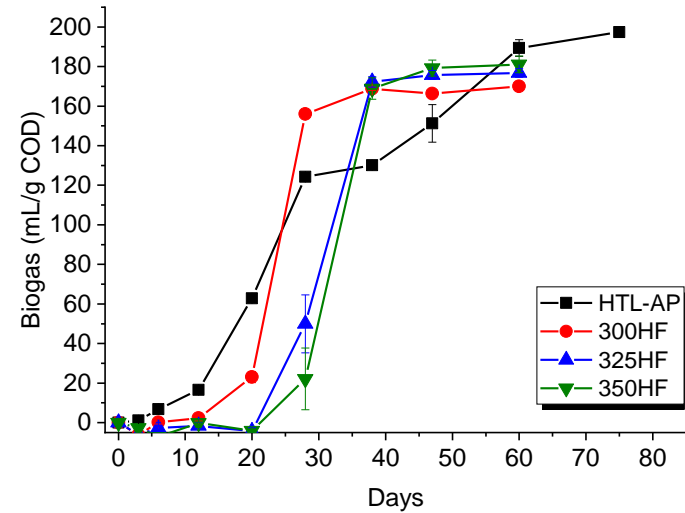


# BIOLOGICAL TREATMENT OF WO WATER

- conversion of ammonia to N-NOx inhibition
- constant v/v dilution



Biomethane potential tests



Sample	B (mL CH <sub>4</sub> /g COD)	Rmax (mL CH <sub>4</sub> /g COD/day)	λ (day)	R <sup>2</sup>
HTL	164	4.2724	9.7128	0.982121
300HF	154	21.9947	19.5267	0.999512
325HF	165	20.1613	25.4737	0.999693
350HF	164	19.6198	27.2293	0.999648

# HR MS ANALYSIS

## Analysis by:

**Jhonattas de Carvalho Carregosa & Alberto Wisniewski Jr**  
(PEB), Department of Chemistry, Federal University of Sergipe,  
Brazil

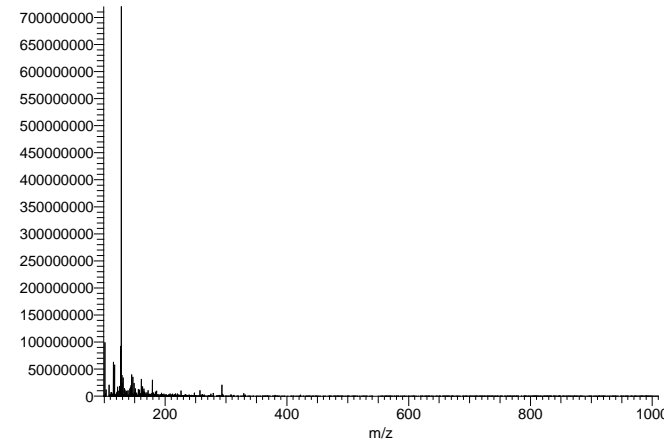
Samples	(-)-HESI		(+) -HESI	
	Number of ions S/N>=3	% of ions reduction	Number of ions S/N>=3	% of ions reduction
HTL-AP	2254	-	2833	-
300-HF	1927	15	1517	46
300-MF	1553	31	1399	51
300-LF	1303	42	1174	59
325-HF	1827	19	1743	38
325-MF	1488	34	1575	44
325-LF	1420	37	1179	58
350-HF	1823	19	1849	35
350-MF1	1259	44	1288	55
350-MF2	1389	38	1656	42
350-LF	837	63	973	66

- Up 65 % reduction in number of ions
- Average Mw reduced from ~180 to ~170 but increase at high severities
- Appearance of new high Mw compounds

## HTL process water (-)

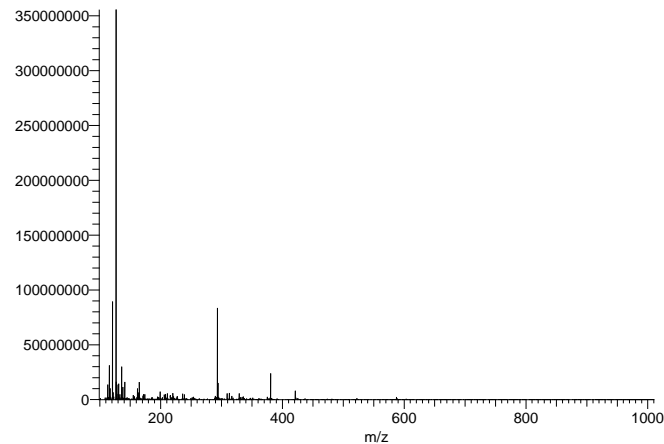
HTL-AP- HESI (-) #140-271 RT: 4.72-5.86 AV: 1:  
T: FTMS - p ESI Full lock ms [100.0000-1000.0000]

7.20E8



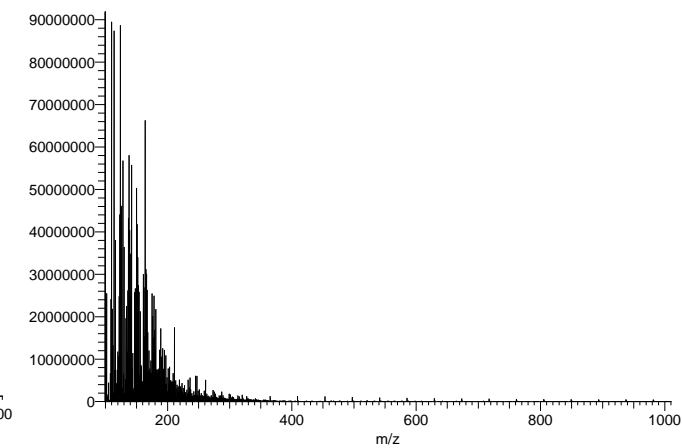
WO-350-50 - HESI (-) #116-206 RT: 4.26-5.04 A  
T: FTMS - p ESI Full lock ms [100.0000-1000.0000] : 3.55E8

## WO water 350°C (-)



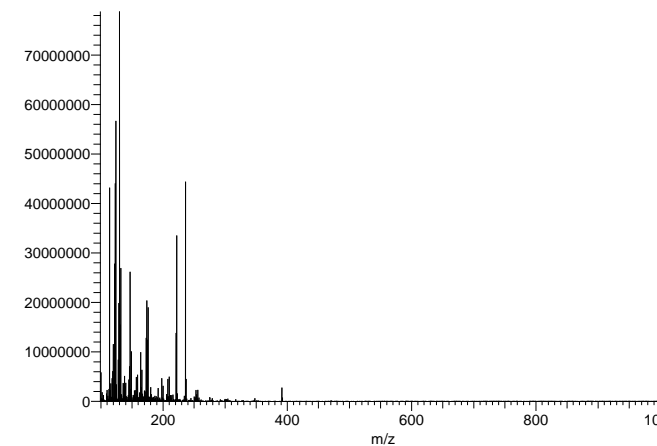
## HTL process water (+)

HTL-AP (HESI) + #146-228 RT: 3.63-4.40 AV: 8:  
T: FTMS + p ESI Full lock ms [100.0000-1000.0000] 18E7



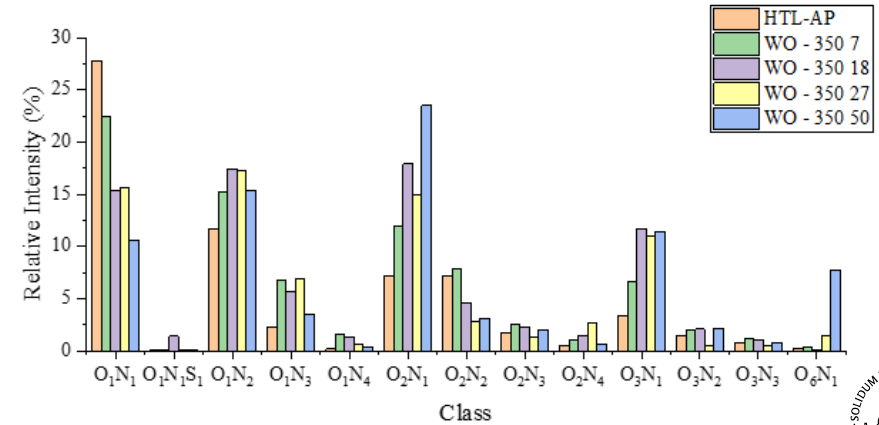
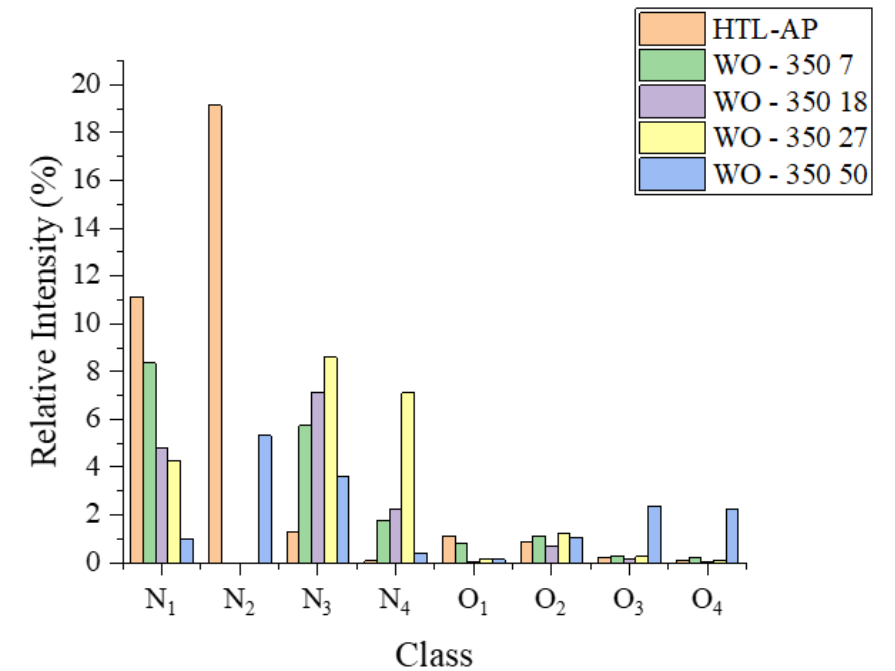
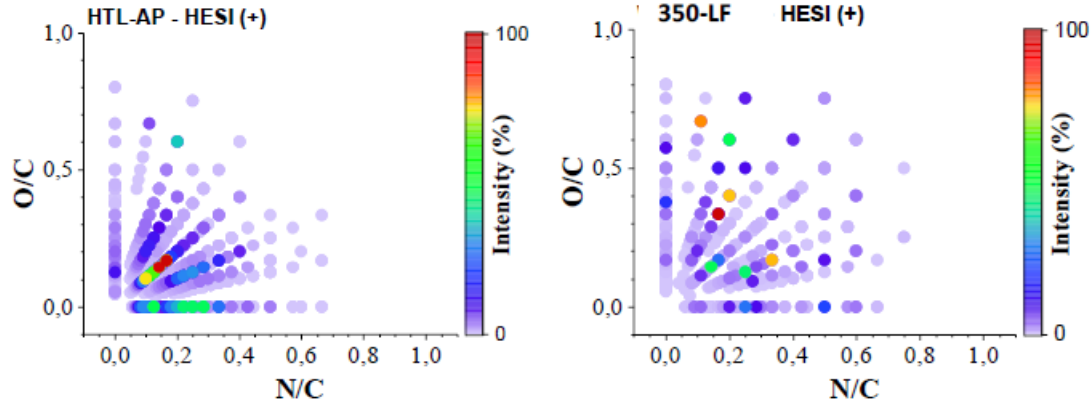
WO-350-50 - HESI(+) #223-355 RT: 9.92-11.07  
T: FTMS + p ESI Full lock ms [100.0000-1000.0000] NL: 7.87E7

## WO water 350°C (+)



# HR MS ANALYSIS

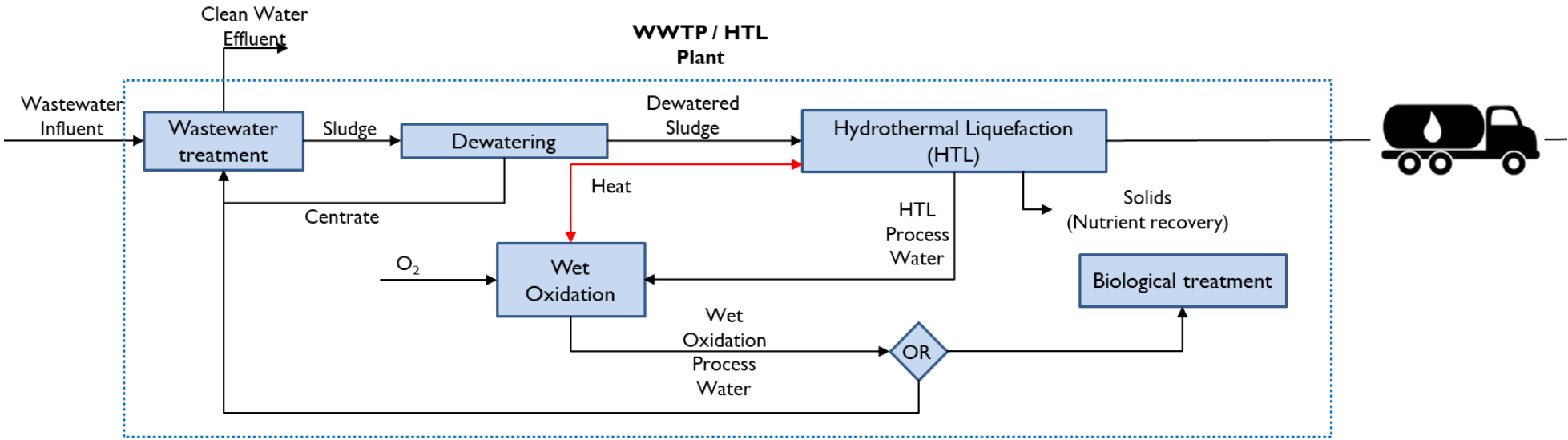
- N1 and N2 degraded
- O1N1, O1N2, O2N1 etc are more recalcitrant



## Analysis by:

Jhonattas de Carvalho Carregosa & Alberto Wisniewski Jr (PEB), Department of Chemistry, Federal University of Sergipe, Brazil

# WET OXIDATION

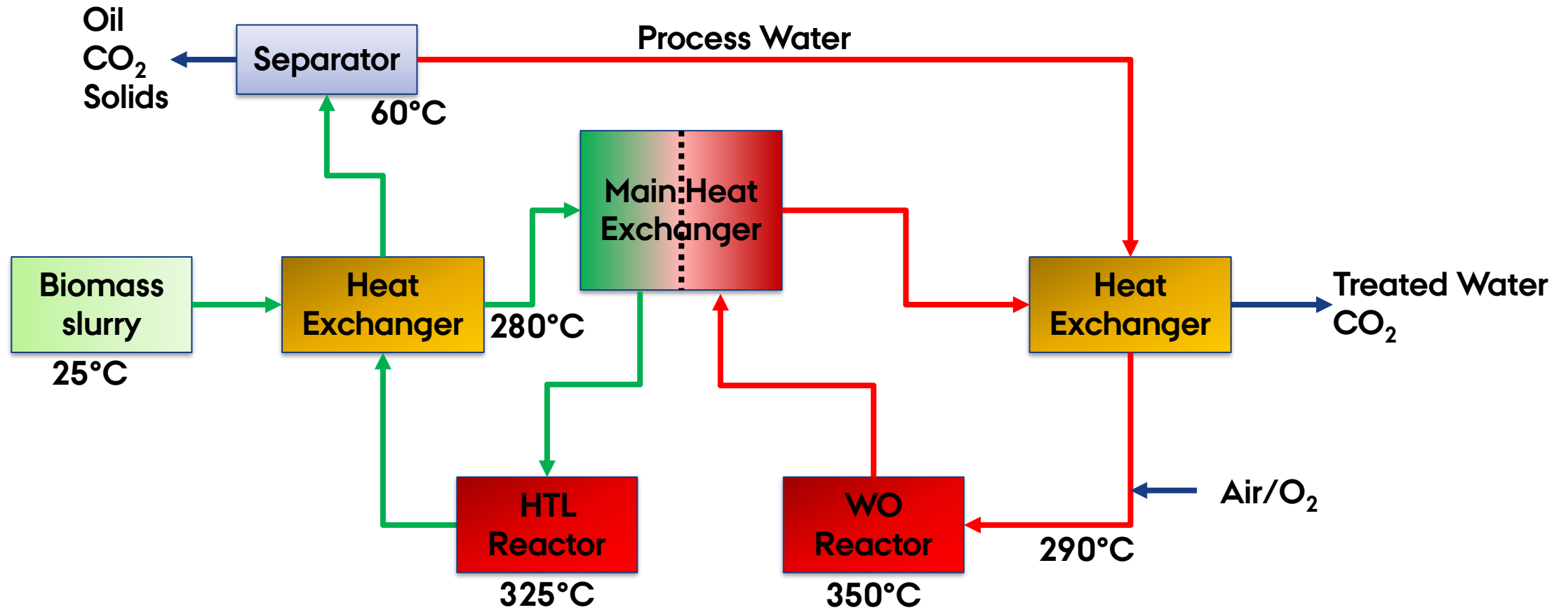


$$ER = COD_{removed}(mol) * 435 \left( \frac{kJ}{mol} O_2 \right)$$

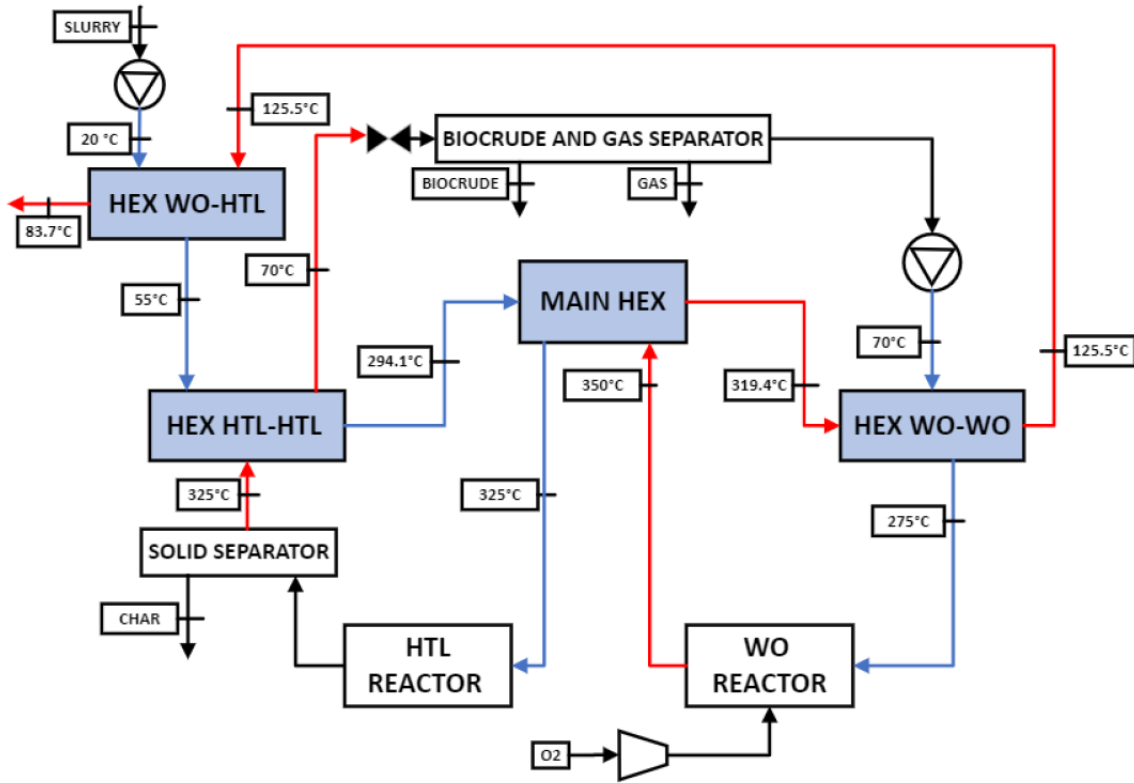
H. Debellefontaine, J.N. Foussard / Waste Management 20 (2000) 15–25



# HTL-WO INTEGRATION



# HTL-WO INTEGRATION



## 3.6 Network setup

The considered fluxes, based on the results of the batch experimental test 325°C/20min, for the exchanger network are reported in the following Table 3.7:

Table 3.7 - Initial flux subdivision based on the 325°C/20min HTL batch experiments and literature data for WO.

Flux name	$T_{in}$ [°C]	$T_{out}$ [°C]	$\dot{m}$ [ $\frac{kg}{s}$ ]	$C_p^*$ [ $\frac{kJ}{kg \cdot K}$ ]	$\dot{m} C_p^*$ [ $\frac{kW}{K}$ ]
Slurry Heating	20	200	1	4.26	4.26
	200	250	1	4.54	4.54
	250	300	1	5	5.00
	300	315	1	5.5	5.50
	315	325	1	5.86	5.86
Aqueous Phase Heating	70	200	0.835	4.27	3.57
	200	250	0.835	4.54	3.79
	250	275	0.835	4.87	4.07
Slurry cooling	325	315	0.91	5.86	5.32
	315	300	0.91	5.5	4.99
	300	250	0.91	5	4.54
	250	200	0.91	4.54	4.12
	200	70	0.91	4.27	3.88
Aqueous phase cooling	350	345	0.885	7.76	6.48
	345	330	0.885	6.84	5.71
	330	315	0.885	5.98	4.99
	315	300	0.885	5.50	4.59
	300	250	0.885	5.00	4.18
	250	200	0.885	4.54	3.79
	200	20	0.885	4.27	3.57

From Table 3.7 is evaluated the total heating and cooling demand:

$$\phi_{max,heatdem.} = 2141 \text{ kW}$$

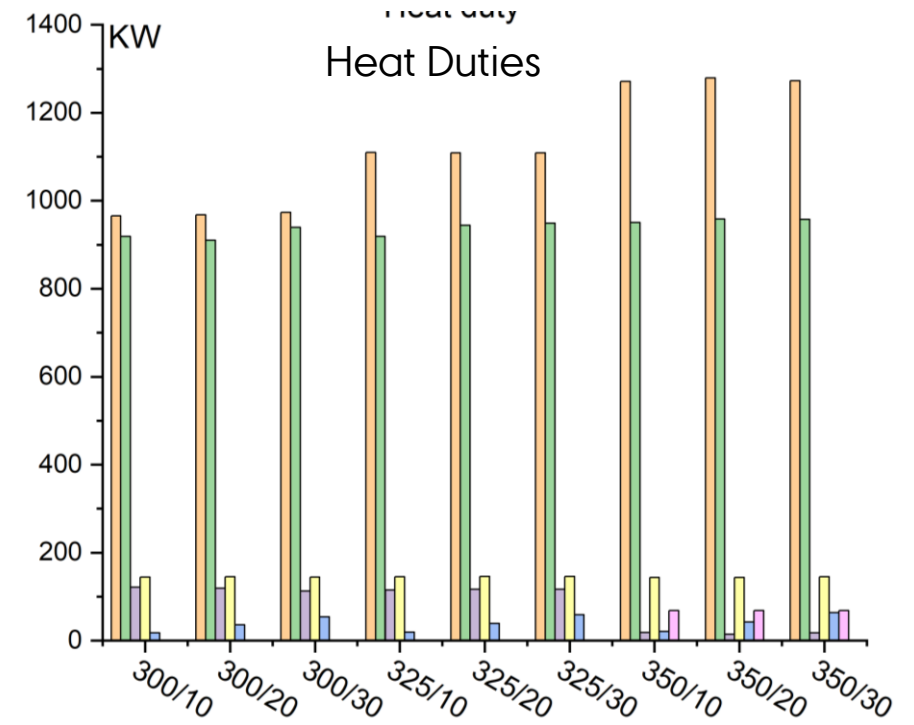
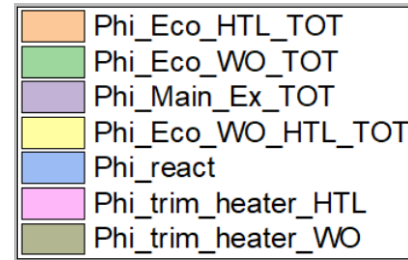
$$\phi_{max,coolingdem.} = 2368 \text{ kW}$$

- Modelling only found 160 KJ/mol vs theoretical 435 KJ/mol O<sub>2</sub>
- Only 55% COD removal achieved in ASPEN model vs ~80% experimental
- Autothermicity confirmed in ASPEN Energy analyser

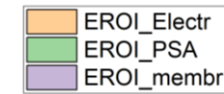
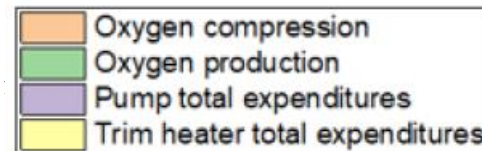
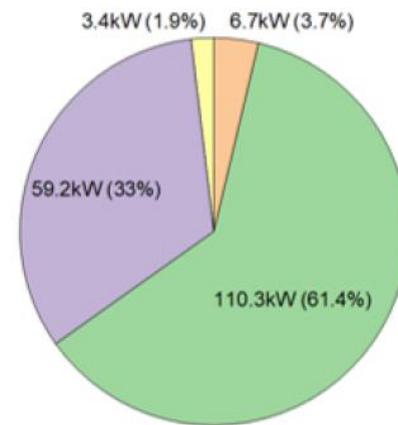


# HTL-WO INTEGRATION

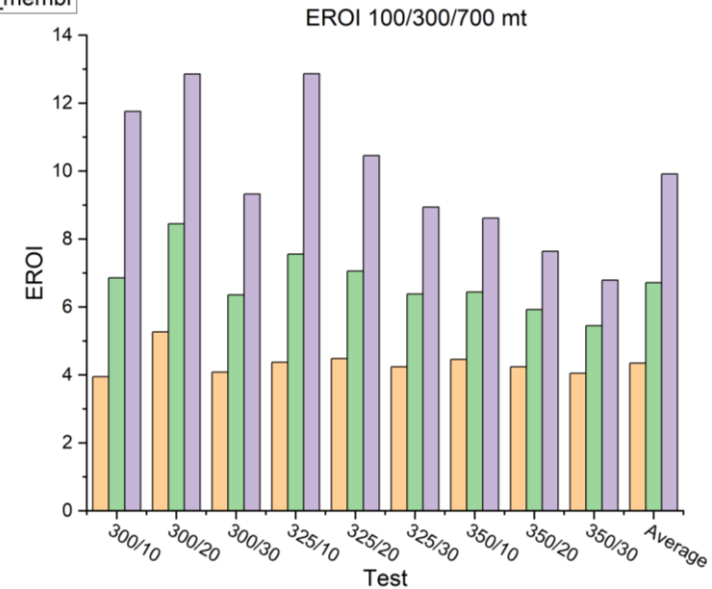
- As long as HTL is not run at 350C and WO at 350C there is no heat demand for HTL
- Pumping and oxygen production are highest energy input
- Source of Oxygen very important for overall energy balance



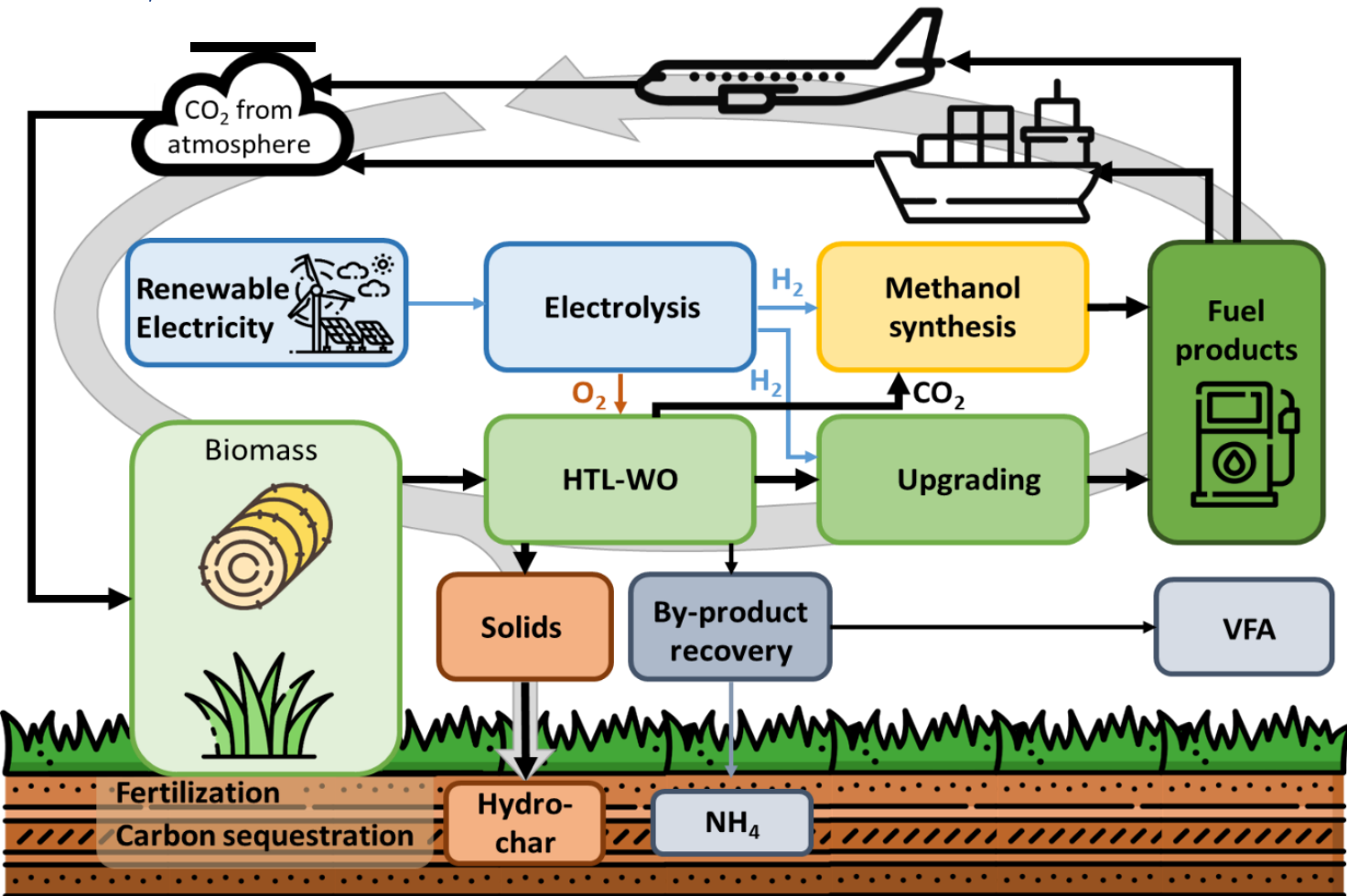
Energy demand system



Energy return on Investment



# CIRCULAIR



*New follow-on HORIZON EU Project*

Objectives:

- *Integrate process water valorization (wetOx)*
- *Producing on spec jet fuel*
- *Carbon negative fuel production*
  - *valorizing all C streams*
  - *Sequestering C in biochar*
- *Autothermal HTL process*
- *Methanol synthesis at Foulum Power-2-X facilities*



<https://project-circulair.eu/>



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# CARBON BALANCE

# CIRCULAIR

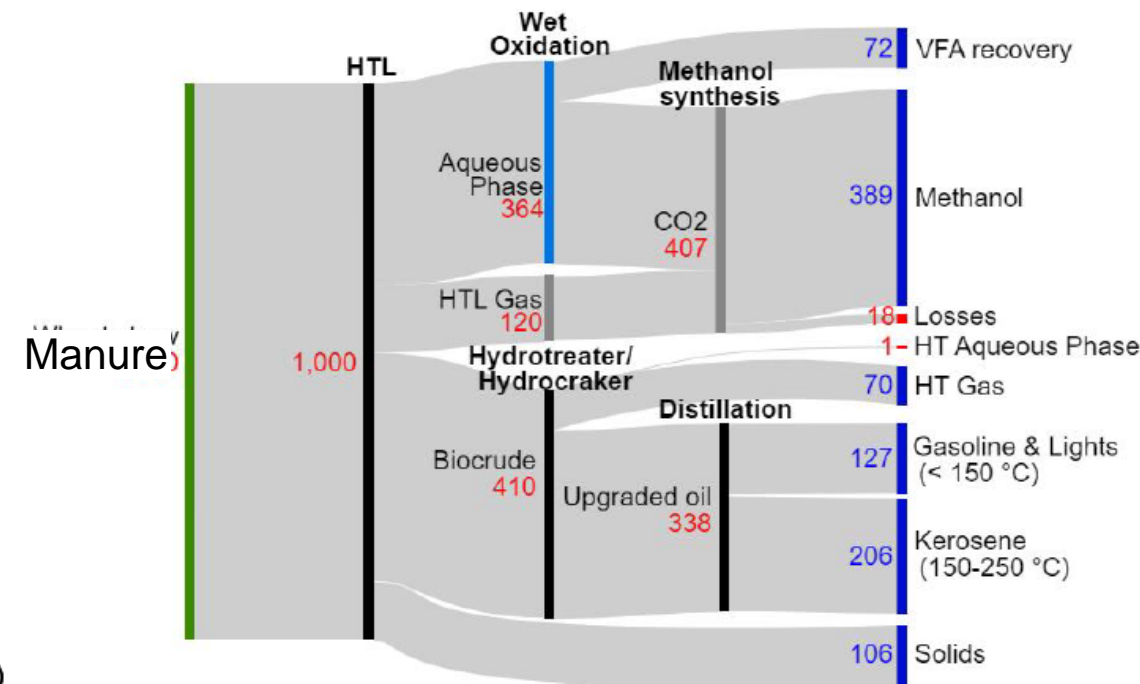
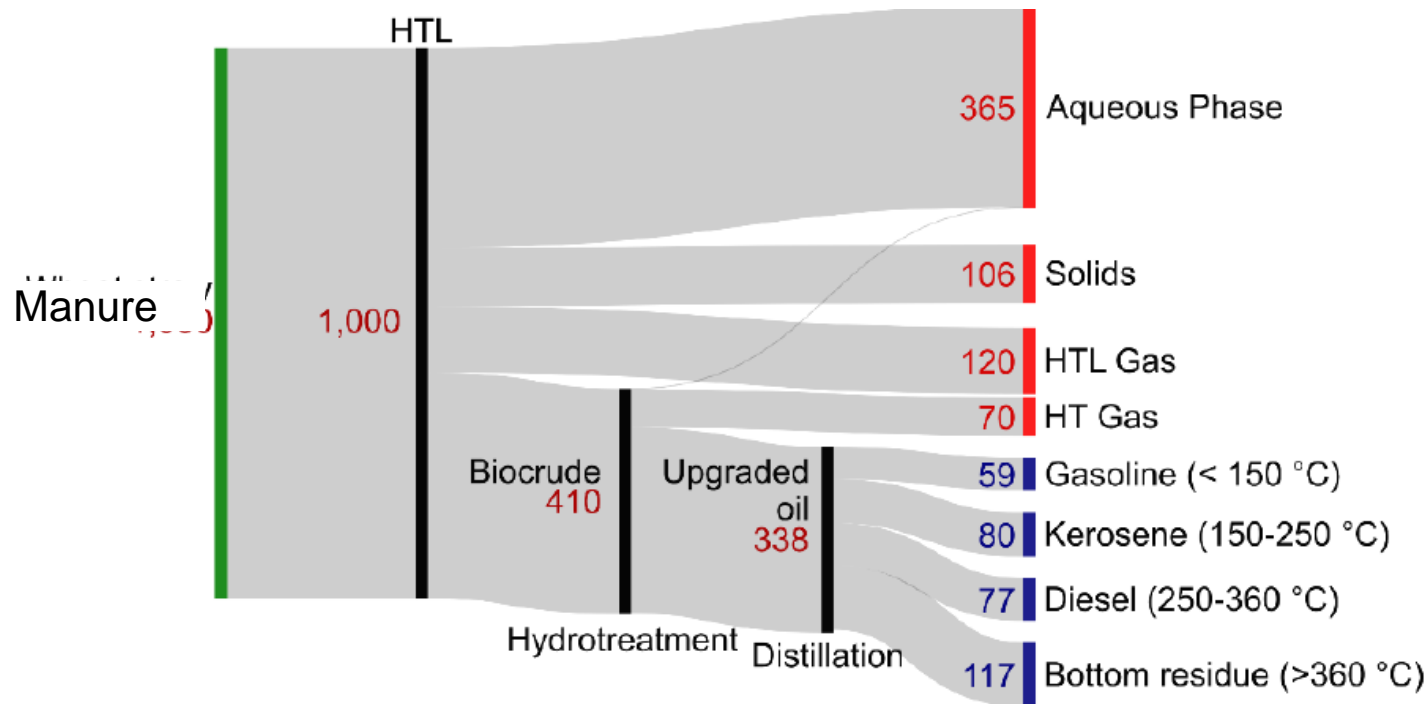
## State of the art:

50% carbon lost to gas and process water

## Ambition:

>90 % carbon utilisation

~20% jet fuel yield



<https://doi.org/10.1016/j.cej.2022.139636>

# SUMMARY

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- WO is efficient in converting HTL water phase organics
- Autothermal integrated HTL-WO is an attractive concept
- Use of CO<sub>2</sub> for Power-2x and O<sub>2</sub> from electrolyzers looks promising
- Recovery of acetic acid and NH<sub>4</sub> should be investigated
- Final use of post-WO water and toxicity needs to be investigated

# THANK YOU



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